

國立清華大學命題紙

96 學年度 工程與系統科學系 (所) 丙 組、先進光源學程 乙組 及 核子工程與科學研究所 甲組 碩士班入學考試

科目 近代物理 科目代碼 3003、3302、3104 共 1 頁第 1 頁 *請在【答案卷卡】內作答

Useful constants: $c = 3.00 \times 10^8$ m/s, $k_B = 1.38 \times 10^{-23}$ J/K, $h = 6.63 \times 10^{-34}$ J·s, $e = 1.60 \times 10^{-19}$ C,
 $m_e = 9.11 \times 10^{-31}$ Kg.

- (12%) A metal surface is irradiated with monochromatic light of variable wavelength. Above a wavelength of 5400 \AA , no photoelectrons are emitted from the surface. With an unknown wavelength a stopping potential of 3 V is necessary to eliminate the photoelectric current. What is the unknown wavelength?
- (15%) X-ray photons of wavelength 0.712 \AA undergo Compton collision in carbon. What is the wavelength change of the line scattered at 90° if the scattering particle is
 - an outer electron ?
 - the whole carbon atom ?
- (15%) Find the deBroglie wavelength of the waves associated with an electron that has been accelerated from rest through a potential difference of
 - 100 V.
 - 800,000 V.
- (13%) Measurements of the mass of a subatomic particle yielded a mass distribution which is centered at $1230 \text{ MeV}/c^2$, and with full width at half maximum about $110 \text{ MeV}/c^2$. Estimate the lifetime of the particle.
- (15%) Determine the mass of a free particle whose wave function is the plane wave

$$\Psi(x, t) = A e^{i(2.5 \times 10^{11} x - 2.1 \times 10^{13} t)},$$

where distance (x) is in meters and time (t) in seconds.

- (15%) One of the excited states of the hydrogen atom has the wavefunction

$$\psi = A r e^{-r/2a} \cos \theta,$$

where a is the Bohr radius.

- Find the normalization constant A .
 - Find the most probable value of r .
- (15%)
 - Write all possible total angular momentum states (j, m_j) for a $3d$ electron in a hydrogen.
 - Assume that the spin-orbit interaction is not overwhelmed by an external magnetic field, what is the minimum angle that the total angular momentum vector \vec{J} may make with the z -axis for the $3d$ electron.
 - What are possible angles between vectors \vec{L} and \vec{S} for the $3d$ electron due to the spin-orbit interaction.